

## Számolási feladatok

### Redoxireakciók

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### 1. Oxidációs számok (válogatva)

- HNO<sub>3</sub>: +1;+5;-2
- H<sub>2</sub>SO<sub>4</sub>: +1; +6;-2
- Fe<sub>2</sub>O<sub>3</sub>: +3;-2
- H<sub>2</sub>O<sub>2</sub>: +1;-1
- KMnO<sub>4</sub>: +1;+7;-2

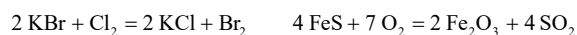
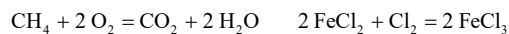
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### 2. Melyek redoxi egyenletek?

- a. SO<sub>2</sub>
- c. Na
- e. H<sub>2</sub>S
- f. diszproporció
- g. SnCl<sub>2</sub>

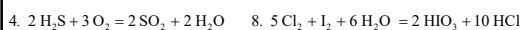
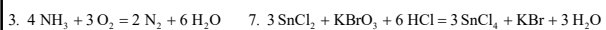
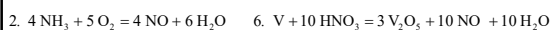
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### 3. Egyenletek



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### 4. Egyenletrendezés



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